

Additional Materials

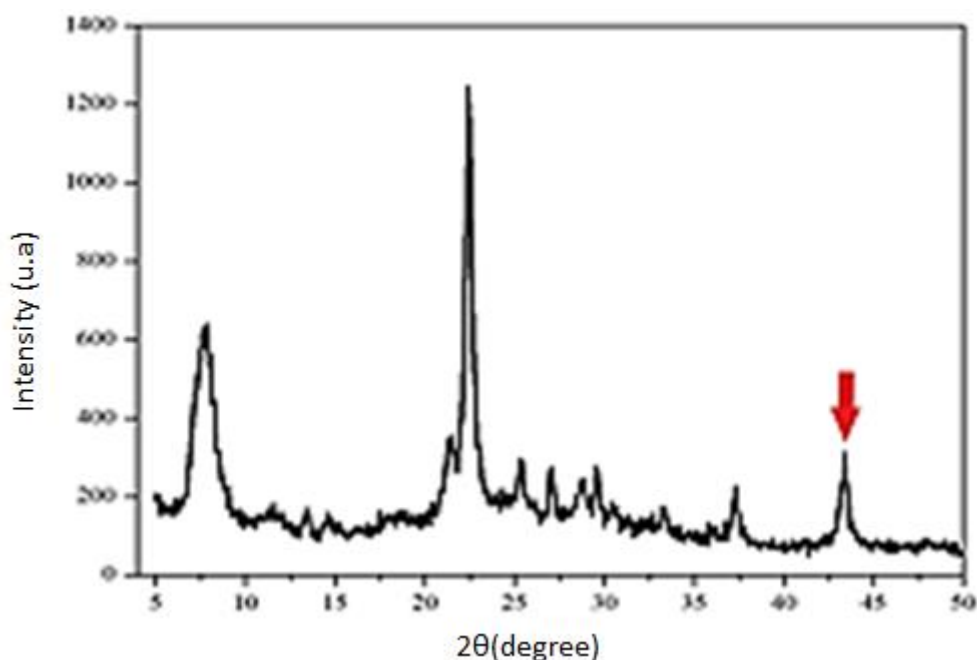


Figure S1 X-ray diffractogram of zeolite beta impregnated with 5% nickel.

Table S1 Silicon, aluminum and nickel contents and specific surface areas and porosity of the catalysts. B = beta zeolite; B3Ni = beta zeolite impregnated with 3% nickel and B5Ni = beta zeolite impregnated with 5% nickel and SU = grape residues without catalyst.

	B	B3Ni	B5Ni	SU
Silicon content (%)	37.8	39.8	35.3	-
Aluminum content (%)	2.85	2.53	2.22	-
Si/Al ratio	13	15	15	-
Nickel content (%)	-	3.56	6.83	-
Sg (m²/g)	459	500	487	-
Vmicro (t-plot) (cm³/g)	0.12	0.16	0.15	-
Vmeso (BJH) (cm³/g)	0.24	0.21	0.18	-
D (BJH) (nm)	6.6	7.9	7.8	-

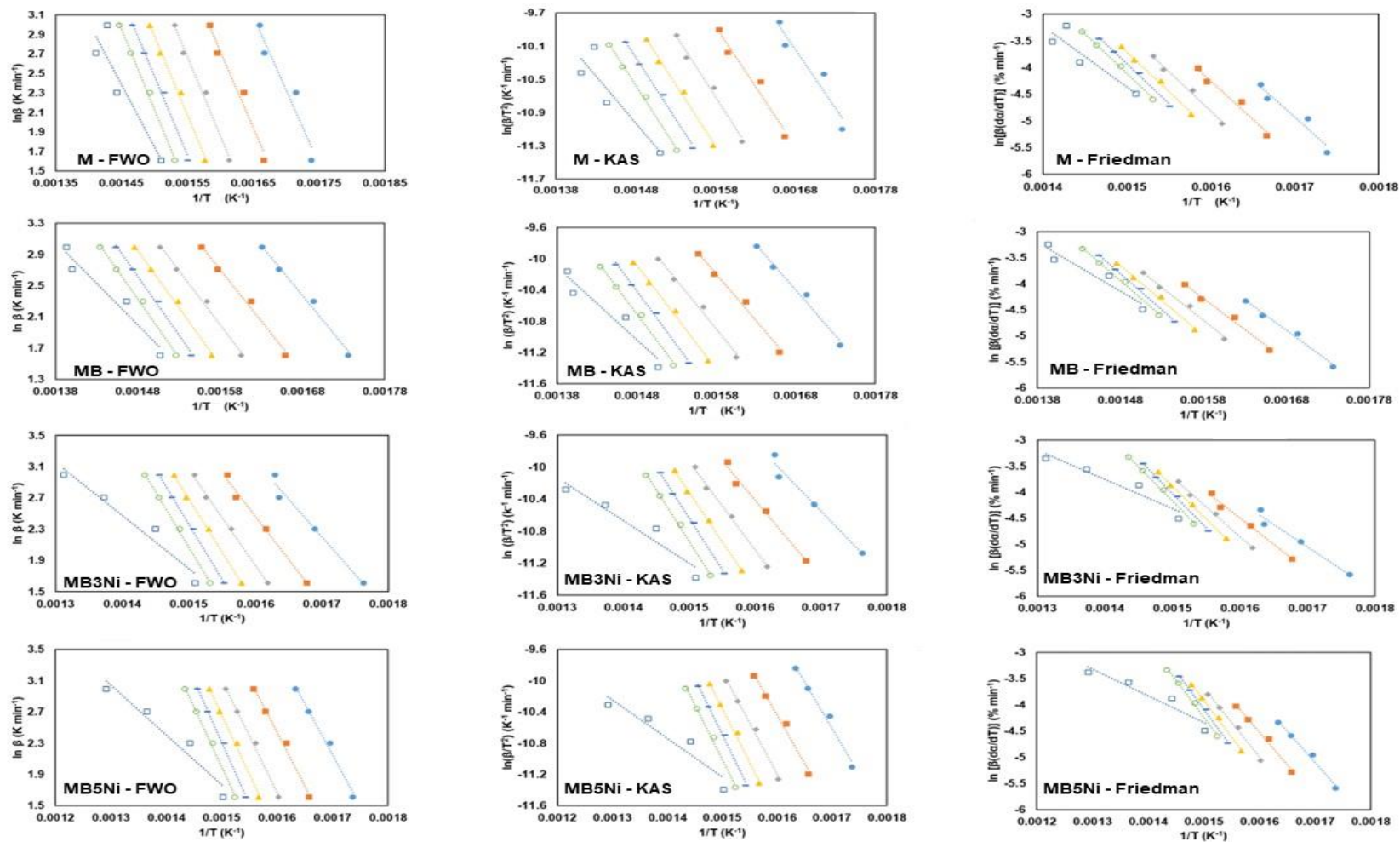


Figure S2 Kinetic curves at different conversion degrees for FWO, KAS and Friedman methods for MDF catalytic and non-catalytic pyrolysis. The lines presented are associated to conversion fraction (a): (-)0.2; (-)0.3; (-)0.4; (-)0.5; (-)0.6; (-)0.7 and (-)0.8.

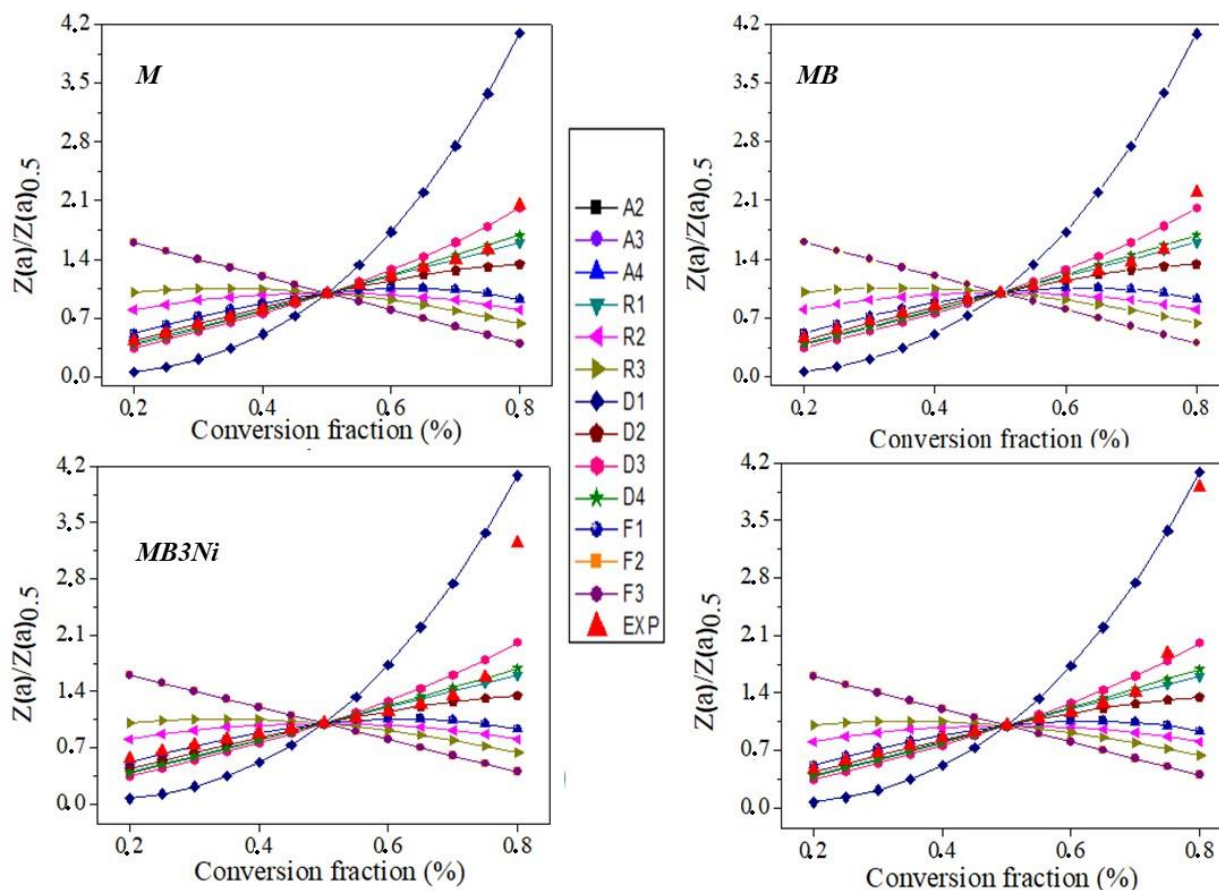


Figure S3 Master curves and experimental data obtained using the Criado method for M, MB, MB3Ni and MB5Ni.

Table S2 Pre-exponential factor (A) for MDF pyrolysis over the catalysts calculated by Friedman (F), Kissinger-Akira-Sunose (KAS) and Flynn-Wall-Ozawa (FWO) isoconversional methods. The results were presented in \log_{10} to facilitate data manipulation.

Log ₁₀ Pre-exponential factor (Log ₁₀ A) – s ⁻¹						
α	M			MB		
	FWO	KAS	Friedman	FWO	KAS	Friedman
0.2	12.71	6.29	8.19	9.95	3.37	5.68
0.3	12.67	6.24	8.13	10.15	3.57	5.93
0.4	12.75	6.31	8.22	10.54	3.96	6.33
0.5	12.65	6.20	8.11	11.04	4.49	6.86
0.6	12.52	6.06	7.99	11.33	4.79	7.19
0.7	12.51	6.03	7.96	11.33	4.77	7.21
0.8	10.19	3.47	5.54	8.72	1.83	4.48
α	MB3Ni			MB5Ni		
	FWO	KAS	Friedman	FWO	KAS	Friedman
0.2	7.76	0.93	3.54	10.20	3.64	5.95
0.3	8.75	1.99	4.59	10.44	3.87	6.21

0.4	9.59	2.90	5.49	10.88	4.33	6.69
0.5	10.36	3.75	6.33	11.46	4.95	7.31
0.6	10.82	4.23	6.83	11.79	5.30	7.67
0.7	10.84	4.23	6.85	11.50	4.97	7.39
0.8	6.39	-0.96	2.00	6.13	-1.30	1.64

Table S3 Thermodynamic parameters for evaluated catalytic and non-catalytic systems for each conversion fraction studied.

Thermodynamic parameters						
α	M			MB		
	ΔG	ΔH	ΔS	ΔG	ΔH	ΔS
0.2	179.25	110.79	-0.10	192.42	90.84	-0.15
0.3	181.43	117.80	-0.10	191.00	92.62	-0.15
0.4	181.94	119.39	-0.09	187.52	94.27	-0.14
0.5	185.28	121.36	-0.10	182.87	96.47	-0.13
0.6	188.77	123.38	-0.10	181.60	99.40	-0.12
0.7	189.58	123.77	-0.10	184.71	102.73	-0.12
0.8	220.58	124.03	-0.15	222.90	105.78	-0.17
α	MB3Ni			MB5Ni		
	ΔG	ΔH	ΔS	ΔG	ΔH	ΔS
0.2	195.45	71.86	-0.18	191.85	99.23	-0.14
0.3	188.36	78.31	-0.16	192.39	103.20	-0.13
0.4	182.11	83.65	-0.15	192.54	109.46	-0.12
0.5	176.42	88.73	-0.13	192.07	117.04	-0.11
0.6	175.64	94.40	-0.12	191.41	120.92	-0.10
0.7	180.80	99.76	-0.12	190.81	116.80	-0.11
0.8	247.76	104.33	-0.21	190.33	42.36	-0.22